

## RECENT SYNTHETIC APPROACHES TO OXADIAZOLE AND THIADIAZOLE DERIVATIVES AS BIOLOGICALLY POTENT ANTI INFLAMMATORY AGENTS

Aparna P.<sup>1\*</sup>, Biju C. R.<sup>2</sup>, Arun Lal V. B.<sup>3</sup>, Princy C.<sup>4</sup>, Ayisha Nitha P.<sup>5</sup>, Shalima N. K.<sup>6</sup> and Babu G.<sup>7</sup>

<sup>1</sup>Lecturer, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

<sup>2</sup>Vice Principal, Department of chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

<sup>3</sup>Professor, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

<sup>4</sup>Lecturer, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra.

<sup>5</sup>Lecturer, Department of chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

<sup>6</sup>Associate Professor, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

<sup>7</sup>Principal, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.



\*Corresponding Author: Aparna P.

Lecturer, Department of Chemistry, Devaki Amma Memorial College of Pharmacy, Chelembra, Malappuram, Kerala, India.

Article Received on 22/09/2024

Article Revised on 12/10/2024

Article Accepted on 02/11/2024

### ABSTRACT

This research focused on clubbing different heterocyclic molecules having diverse biological action to produce potentially active derivatives of oxadiazole and thiadiazole. In the present study we have revealed a new synthetic approach for synthesizing novel oxadiazole and thiadiazole derivatives. Synthesis of 1, 3, 4 – Oxadiazole and Thiadiazole were carried out by the reaction between chalcone and isonicotinic acid hydrazide, followed by cyclization with propyl phosphonic anhydride for Oxadiazole and Hurd Mori reaction for Thiadiazole. Purity of the compounds ascertained consistency by TLC and melting point determination. The structure of newly synthesized compounds was characterized by IR, HNMR, MASS Spectral analysis. By comparing the results of all compounds, we have reached in a conclusion that 3a5 and 3b5 can be considered as potent anti-inflammatory agents.

**KEYWORDS:** Chalcone, Oxadiazole, Thiadiazole, anti-inflammatory.

### INTRODUCTION

Chalcones are 1, 3-diphenyl-2-propene-1-one, consist of two aromatic rings are interconnected by highly electrophilic three carbon  $\alpha$ , and  $\beta$ -unsaturated carbonyl system that assumes linear or nearly planar structure. They containing keto ethylinic group (-CO-CH=CH-). Chalcones possess conjugated double bonds and completely delocalized  $\pi$ - electron system on both benzene rings. The presence of a reactive and unsaturated keto function in chalcones is found to be responsible for their antimicrobial activity.

Oxadiazole is a five membered heterocyclic compound containing one oxygen atom and two nitrogen atoms. It is considered to be derived from furan by substitution of two -CH= groups with two pyridine type nitrogen (-N=).

Oxadiazole and their derivatives had been reported to exhibit several biological activities like, anti-inflammatory, antifungal, antibacterial, anticonvulsant, and mono amino oxidase inhibition. It occurs in various isomeric forms like, 1, 2, 3-oxadiazole, 1, 2, 5-oxadiazole, 1, 2, 4-oxadiazole and 1, 3, 4-oxadiazole. However, 1, 3, 4 and 1, 2, 4-oxadiazole are better known, and more widely studied by researchers, because of their important chemical and biological properties. In chemistry, thiadiazoles are a sub family of azole compounds. Structurally they are five membered heterocyclic compounds containing two nitrogen and a sulfur atom, and two double bonds, to give an aromatic ring. Four possible structures exist depending on the relative positions of the heteroatoms such as 1, 2, 3- thiadiazole, 1, 2, 4- thiadiazole, 1, 2, 5-

thiadiazole and 1,3,4- thiadiazole. It exhibits a wide variety of pharmacological activities such as anticancer, antitubercular, antibacterial, antifungal, antimicrobial, antiinflammatory, analgesic, anticonvulsant, diuretic and antisecretory activity. This study aims to synthesize potentially active novel derivatives of oxadiazole and thiadiazole by incorporating different heterocyclic moieties and substituting different functional groups. Synergism may occur due to the incorporation of two or more moiety that results in the formation of highly potent compound.

## MATERIALS AND METHOD

### Synthesis and Characterization

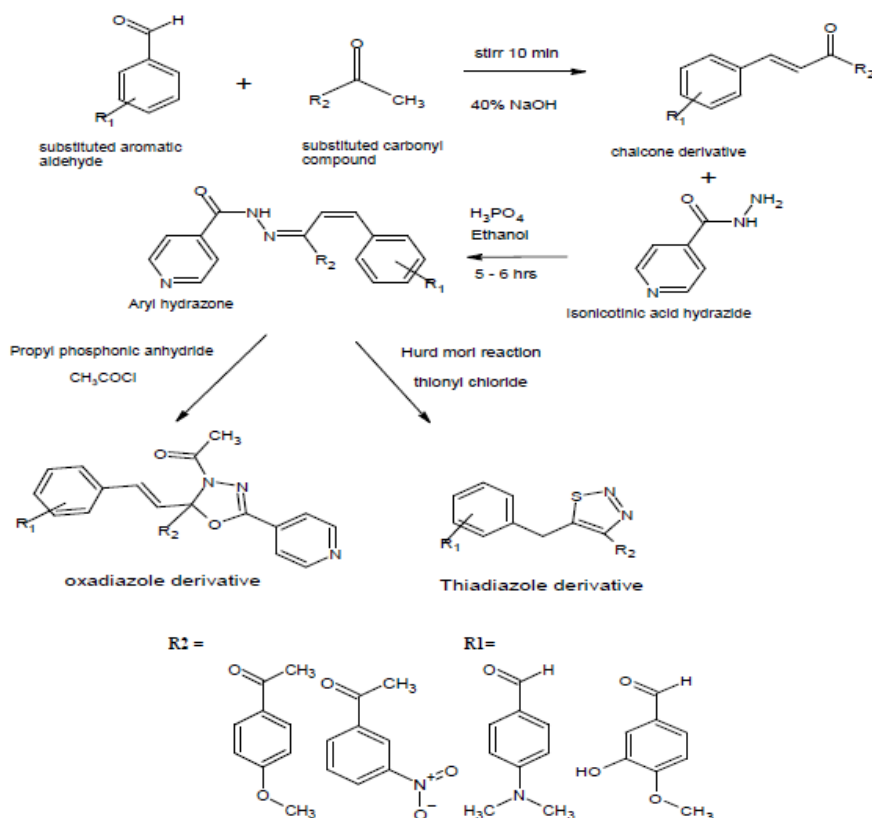
All the chemicals and reagents used in this research work were analytical or practical grade. The compound procured were purified and dried using standard methods before use, wherever necessary. Melting point of the

synthesized compounds were determined by open capillary method, Infra-Red spectra of the synthesized compounds are recorded using JASCO FT-IR spectrophotometer. Proton NMR spectra of synthesized compounds are recorded in D<sub>2</sub>O on Bruker ultra-shield DPX 400 spectrophotometer. Mass spectra of the synthesized compounds are recorded by using LC-MSD Trap-SL 2010 A- Purity of the compound ascertained by TLC over precoated, preactivated glass plates with appropriate solvent systems. Purity of the compounds are confirmed by single spot in TLC and consistency in the R<sub>f</sub> value.

### Pharmacological activity: Anti-inflammatory studies

Antiinflammatory activity of synthesized compounds are evaluated by Carrageenan induced paw oedema in rats from Devaki Amma Memorial college of Pharmacy, Chelembra, Malappuram.

## EXPERIMENTAL SECTION



### General Procedure for the preparation of chalcone derivatives

Dissolve 0.01 mol benzaldehyde and 0.01 mol acetophenone in 10 ml 95% ethanol in 25 ml Erlenmeyer flask and equipped with magnetic stirrer bar. 3.5 ml of NaOH solution was added to the reaction flask and stirred for 10 min. Cooled the mixture until the crystal formation was completed. Added 2ml of ice cold ethanol and allow to air dry. Recrystallized from 95% ethanol.

### General procedure for the synthesis of aryl hydrozones

A mixture of isonicotinic acid hydrazide (1.81 mM) and chalcones (1.81 mM) in absolute ethanol 20 ml and catalytic amount of H<sub>3</sub>PO<sub>4</sub> was refluxed for 10 hours. Reaction completion was monitored by thin layer chromatography (TLC). Reaction mixture was then poured on to crushed ice, solid separated was filtered and dried. Crude compound was purified by recrystallization using hot water to get pure aryl hydrozones as colorless solid.

### General procedure for the synthesis of oxadiazole derivatives

A mixture of aryl hydrazones (Schiff's base) derivative, Propyl phosphonic anhydride (T3P) (10 mM) and acetyl chloride (15 volume) was refluxed for 8 hours. Progress of the reaction was monitored by Thin Layer Chromatography. After completion of the reaction, reaction mixture was cooled to room temperature and excess acetyl chloride was removed by the addition of saturated sodium bicarbonate solution to obtain crude compound.<sup>[49]</sup>

### General procedure for synthesis of thiadiazole derivatives

An excess amount of thionyl chloride was stirred at 0°C and the corresponding hydrazones were added in several portions. The mixture were stirred at room temperature until no more hydrogen chloride was produced. The remaining thionyl chloride was evaporated under vacuum and the residue was washed with diethyl ether. Recrystallisation from hot water.<sup>[70]</sup>

### Synthesis of 1-[2-[(E)-2-[4-(dimethylamino)phenyl]ethenyl]-2-(3-nitrophenyl)ethenyl - 5-(pyridin-4-yl)-1,3,4-oxadiazol-3(2H)-yl]ethan-1-one (3a5)

A mixture of aryl hydrazones (Schiff's base) (2a5) derivative, Propyl phosphonic anhydride (T3P) (10 mM) and acetyl chloride (15 volume) was refluxed for 8 hours. Progress of the reaction was monitored by Thin Layer Chromatography. After completion of the reaction, reaction mixture was cooled to room temperature and excess acetyl chloride was removed by the addition of saturated sodium bicarbonate solution to obtain crude compound.

### Synthesis of 1-[2-[E]-2-(4-hydroxy-3-methoxyphenyl)ethenyl]-2-(Methoxyphenyl)ethenyl-5-(pyridin-4-yl)-1,3,4-oxadiazole-3(2H)-yl]ethan-1-one (3a6)

A mixture of aryl hydrazones (Schiff's base) (2a6) derivative, Propyl phosphonic anhydride (T3P) (10 mM) and acetyl chloride (15 volume) was refluxed for 8 hours. Progress of the reaction was monitored by Thin Layer Chromatography. After completion of the reaction, reaction mixture was cooled to room temperature and excess acetyl chloride was removed by

the addition of saturated sodium bicarbonate solution to obtain crude compound.

### Synthetic procedure for N, N-dimethyl-4-[[4-(3-nitrophenyl)-1, 2, 3-thiadiazol-5-yl]methyl]aniline (3b5)

An excess amount of thionyl chloride was stirred at 0°C and the corresponding hydrazones (2a5) were added in several portions. The mixture were stirred at room temperature until no more hydrogen chloride was produced. The remaining thionyl chloride was evaporated under vacuum and the residue was washed with diethyl ether. Recrystallization from hot water.

### Pharmacological evaluation Animals

Albino mice of swiss strains and wistar rat were used for the pharmacological and toxicological studies. The animal's experimental protocol has been approved by our Institutional Animal Ethics Committee (IAEC) registration no: DAMCOP/IAEC/045.

### Acute toxicity study

The acute oral toxicity study was carried out on Swiss Albino mice as per the guidelines No: 423 given by the organization for Economic Co-operations and Development (OECD 423, 1988).

### Antiinflammatory study

#### Acute Antiinflammatory study by Carrageenan induced paw edema in rat

This method is the most commonly used method for the evaluation of antiinflammatory drugs.

Group1: Control (normal saline)

Group2: Standard group (Indomethacin 10mg/kg body weight)

Group3: Oxadiazole derivative (lower dose+ higher dose)

Group4: Thiadiazole derivative (lower dose+higher dose)<sup>83</sup>

### RESULTS

Table 1 Preliminary characterization of synthesized compounds. Table 2: Spectral analysis of synthesized compounds. Figure 1: Histogram showing Antiinflammatory effect of derivative 3a5 by carrageenan induced paw edema. Figure 2: Histogram showing Antiinflammatory effect of derivative 3a5 by carrageenan induced paw edema.

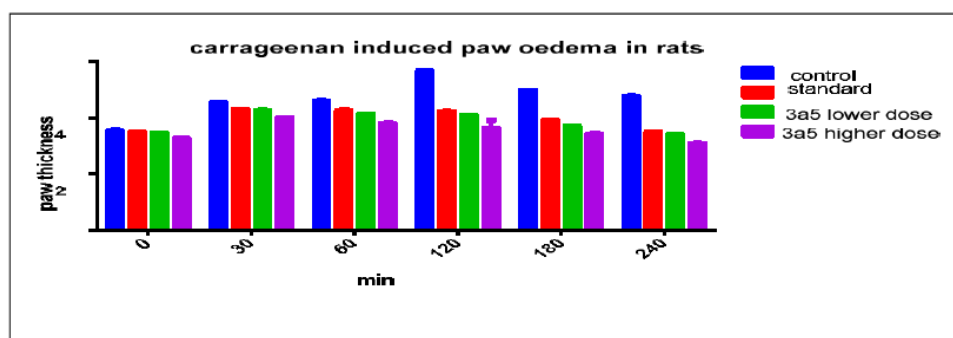


Figure 1: Histogram showing Antiinflammatory effect of derivative 3a5 by carrageenan induced edema.

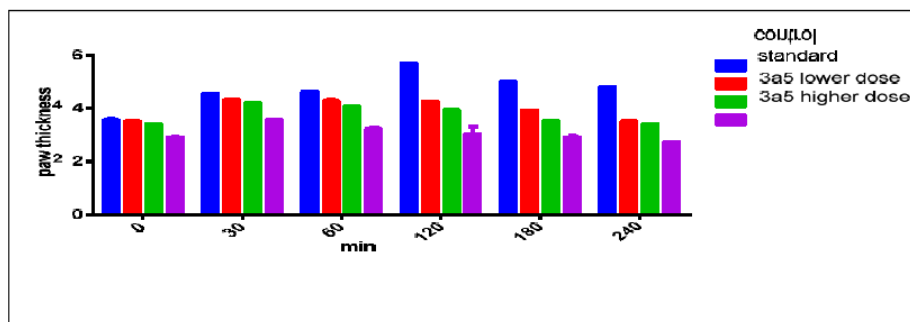


Figure 2: Histogram showing Antiinflammatory effect of derivative 3b5 by induced paw edema.

Table 1: Preliminary characterization of synthesized compounds.

Compound code	Molecular formula	Molecular weight	Melting Point ( $^{\circ}$ C)	Percentage Yield	Rf value
3a5	C <sub>25</sub> H <sub>23</sub> N <sub>5</sub> O <sub>4</sub>	457.49	200	60	0.6
3a6	C <sub>25</sub> H <sub>23</sub> N <sub>3</sub> O <sub>5</sub>	445.48	195-200	65	0.6
3b5	C <sub>17</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> S	340.1	100-130	75	0.7

Table 2: Spectral analysis of synthesized compounds.

Compound code	IR peaks (cm)	<sup>1</sup> H NMR ( $\delta$ value in PPM)	<sup>13</sup> C NMR ( $\delta$ value in PPM)	Mass spectral details
3a5	3020 (Ar C-H str); 2934 (CH <sub>3</sub> str); 1700 (C=O str); 1342 (C-N str); 1099 (C-O str); 1527 (NO <sub>2</sub> str)	6.54-7.12 (D, 4H, CH-Benzene), 2.02-2.86 (S, 9H, CH <sub>3</sub> ), 8.02-8.84 (D, 4H, CH-pyridine), 7.45-8.12 (4H, CH-	23.8, 39.83, 40.00, 40.33, 40.331, 82.4, 114.20, 114.21, 119.1, 122.21, 124.1, 124.11, 124.7, 126.30, 127.31,	Base peak atm/z value 411 Molecular ion peak (M+H) atm/z value 457 Molecular weight of the compound is 457.49
		Benzene) 6.34-6.66 (D, 2H, ethylene)	127.311, 129.5, 129.7, 141.6, 148.2, 148.8, 149.5, 149.51, 155, 168.6.	
3a6	3655 (O-H str); 3011 (Ar C-H str); 2825 (O-CH <sub>3</sub> str); 1780 (C=O str); 1599 (CH=CH alkene str); 1300 (C=N str); 1030.77 (C-O str)	6.57-6.69 (D, 3H, CH-Benzene), 2.02-3.73 (S, 9H, CH <sub>3</sub> ), 6.70-7.08 (D, 4H, CH-Benzene), 8.02-8.34 (D, 4H, CH-Pyridine), 5.0 (D, 1H, -OH), 6.34-6.66 (D, 2H, Ethylene)	23.8, 55.9, 56.2, 83.4, 112, 114.1, 114.2, 116.8, 120.1, 124.0, 124.1, 126.5, 128, 128.1, 128.2, 129.7, 133, 133.1, 138, 149.5, 149.51, 151.3, 155, 158.7, 168.	Base peak atm/z value 150 Molecular ion peak (M+H) atm/z value 445 Molecular weight of the compound is 445.48
3b5	3100 (Ar C-H str); 2945 (CH <sub>3</sub> str); 1588 (NO <sub>2</sub> str); 1501 (N=N str); 1431 (C=C str); 1365 (C-N str); 800 (C-S str)	6.47-6.88 (D, 4H, CH-Benzene), 3.81 (D, 2H, CH <sub>2</sub> ), 7.58-8.41 (D, 4H, CH-Benzene), 2.85-2.851 (S, 6H, CH <sub>3</sub> .)	28.9, 40.3, 40.31, 40.32, 114.0, 114.01, 121.1, 121.11, 125.8, 128.9, 130.0, 130.12, 133.6, 134.0, 146.6, 148.9, 158.2.	Base peak atm/z value 329 Molecular ion peak (M+H) atm/z value 340 Molecular weight of the compound is 340.1

## DISCUSSION

Synthesis of oxadiazole and thiadiazole derivatives were carried out by fusing chalcone with isonicotinic acid hydrazide, followed by cyclization for oxadiazole derivatives and, by Hurd Mori reaction for Thiadiazole derivatives. Purity of the compounds were ascertained consistency by TLC and melting point determination. The structure of newly synthesized compounds were characterized by IR, HNMR, MASS Spectral analysis. From the antiinflammatory study by carrageenan induced paw edema method it is revealed that compounds 3a5 and 3b5 have shown better reduction in the carrageenan induced paw edema when compared to standard. The compound 3a5 and 3b5 with a dose of 400mg/kg have shown better antiinflammatory activity than the dose of 200mg/kg.

From these results and observation it was found that in future these newly synthesized 1, 3, 4 – Oxadiazole and 1,2,3 Thiadiazole derivatives can be developed as a lead molecules in antiinflammatory, drug discovery process.

## ACKNOWLEDGEMENT

The author is thankful to the management of Devaki Amma Memorial College of Pharmacy, Malappuram for Providing the Facilities to carry out this research work.

## REFERENCES

1. H.K. Donda, S. D. Faldu. Vital role of aromatic ketone to demonstrate various biological activities : A chalcone. *International Journal of Drug Design and Discovery*, 2013; 4(2): 1031-1049.
2. Hatish Prashar, Anshul Chawla, Anil Kumar Sharma, Rajeev Kharb. Chalcone as a versatile moiety for diverse pharmacological activities. *International Journal of Pharmaceutical Sciences and Research*, 2012; 3(7): 1913-1927.
3. Demetrios, N. Nicolaidis, C. Konstantina, FylaktakidouKonstantinosLitinas. Synthesis and biological evaluation of several coumarin-4-carboxamidoxime and 3-(coumarin- 4-yl)-1, 2, 4-oxadiazole derivatives. *European Journal of Medicinal Chemistry*, 1998; 33: 715-724.
4. Subin Mary Zachariah, Mridula Ramkumar, Namy George, Mohammad Salam Ashif. A Review on Oxadiazole. *Research Journal of Pharmaceutical Biological and Chemical Sciences*, 2015; 6(2): 205-209.
5. Anitha Huliurdurga C., Sreenivasa S., Mohan N., Vivek C., Shivaraja G. Synthesis, characterisation, *invivo*, *invitro* and *insilico* studies of some novel substituted 1,3,4 oxadiazole-3(2H)yl –ethanone derivatives. *World Journal of Pharmacy and Pharmaceutical sciences*, 2017; 6: 1897-1917.
6. Ramaiyan manikannan, Masilamani shanmugaraja, Seetharaman manojveer and Shanmugam muthusubramanian. Synthesis and characterization of 5- heteroarylsulfanyl-4-aryl 1, 2, 3-selena/thiadiazoles. *Journal of Chemical Sciences*, 2012; 124(2): 463-468.
7. OECD guideline 423.
8. Parmar Namitha, Rawat Mukesh, Kumar Tirath, Evaluation of anti-inflammatory potential of *Kigelia pinnata* leaf extract in wist rats. *Asian Journal of Pharmaceutical and Clinical Research*, 2011; 5(1): 95-97.