

## STABILITY INDICATING UPLC METHOD OPTIMISATION AND VALIDATION OF TRIAMCINOLONE IN SYRUP DOSAGE FORM

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### ABSTRACT

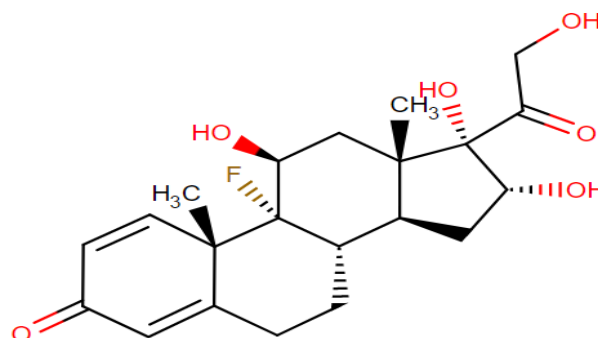
A selective, precise, accurate and stability indicating UPLC method is validated for estimation of Triamcinolone in syrup dosage form. The method employed, with Hypersil BDS C18 (100 mm x 2.1 mm, 1.7  $\mu$ m) column in gradient mode, with mobile phase of Methanol and Acetonitrile in the ratio of 45:55 %v/v. The flow rate was 1.2 ml/min and effluent was monitored at 238nm. Retention time was found to be 5.212 $\pm$ 0.58 min. The method was validated in terms of linearity, accuracy, precision, limit of detection (LOD), limit of quantification (LOQ) etc. in accordance with ICH guidelines. Linear regression analysis data for the calibration plot showed that there was good linear relationship between response and concentration in the range of 8- 40 $\mu$ g/ml respectively. The LOD and LOQ values for were found to be 0.8426( $\mu$ g/ml) and 2.5535( $\mu$ g/ml) respectively. No chromatographic interference from syrup's excipients and degradants were found. The proposed method was successfully used for estimation of Triamcinolone in syrup dosage form.

**KEYWORDS:** Triamcinolone, UPLC, Validation, stability indicating method.

### 1. INTRODUCTION

Triamcinolone, (8S, 9R, 10S, 11S, 13S, 14S, 16R, 17S) -9- fluoro- 11, 16, 17- trihydroxy-17-(2-hydroxyacetyl)-10,13-dimethyl-6,7,8,11,12,14,15,16-octahydrocyclopenta[a]phenanthren-3-one (Fig.1),<sup>[9]</sup> is an intermediate-acting glucocorticoid, as the free alcohol or in esterified form, it is administered orally, intramuscularly, by local injection, by inhalation, or applied topically in the management of various disorders in which corticosteroids are indicated.<sup>[1-4]</sup>

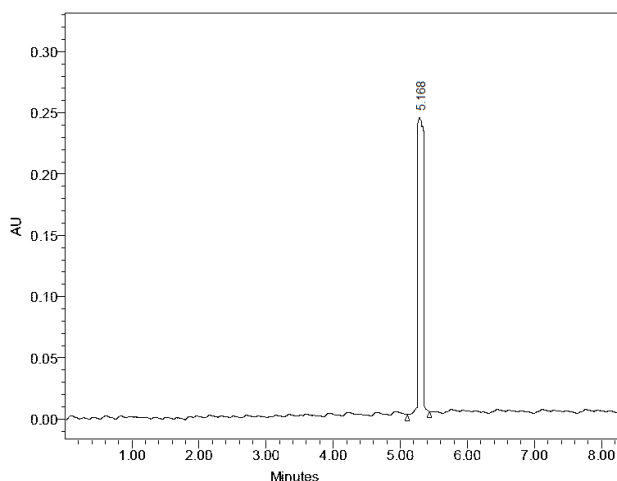
The stability of Triamcinolone is a matter of great concern as it affects the safety and efficacy of the finished syrup product. Stability indicating studies provide data to support identification of possible degradants; degradation pathways and intrinsic stability of the Triamcinolone molecule and validation of stability indicating analytical procedures.



**Fig.1: Molecular Structure of Triamcinolone, (8S, 9R, 10S, 11S, 13S, 14S, 16R, 17S) -9- fluoro- 11, 16, 17- trihydroxy-17-(2-hydroxyacetyl)-10, 13-dimethyl-6, 7, 8, 11, 12, 14, 15, 16 -octahydrocyclopenta[a] phenanthren-3-one<sup>[9]</sup>**

Regulatory agencies recommend the use of stability indicating methods (SIMs) for the analysis of stability samples. This requires stress studies in order to generate the potential related impurities under stressed conditions, method development and validation. With the evident of the International Conference on Harmonization (ICH) guidelines, requirements for the establishment of SIMs have become more clearly mandated.<sup>[1-3]</sup> Environmental

conditions including light, heat and the susceptibility of the drug product towards hydrolysis or oxidation can play an important role in the production of potential impurities.<sup>[10-11]</sup> Stress testing can help identifying degradation products and provide important information about intrinsic stability of the drug product.<sup>[12-14]</sup> Therefore, herein we report the results of stability study of Triamcinolone with the aim of determining the extent of the influence of different stress conditions on the stability of the syrup product.



**Fig. 2: Standard Chromatogram of Triamcinolone, using mobile phase of Methanol and Acetonitrile in the ratio of 45:55 %v/v.**

## 2. EXPERIMENTAL

### Materials

Triamcinolone (99.80 % purity) used as analytical standard was procured from Active Pharma Labs (Hyderabad).

HPLC grade methanol, Acetonitrile (HPLC grade) was purchased from Qualigens fine chemicals, Mumbai, India. Distilled, 0.45  $\mu$ m filtered water used for UPLC quantification and preparation of buffer. Buffers and all other chemicals were analytical grade.

The commercial syrup-dosage (Aristocort - Syr Orl) labeled to contain 0.4mg/ml of Triamcinolone. All chemicals used were of pharmaceutical or special analytical grade.

### Instrumentation

Acquity, Waters UPLC system consisting of a Water 2695 binary gradient pump, an inbuilt auto sampler, a column oven and Water 2996 wavelength absorbance detector (PDA) was employed throughout the analysis.

The data was collected using Empower 2 software. The column used was Hypersil BDS C18 (100 mm x 2.1 mm, 1.7  $\mu$ m). A Band line sonerex sonicator was used for enhancing dissolution of the compounds. A Labindia pH System 362 was used for pH adjustment.

## Chromatographic Conditions

**Table 1: Chromatographic Conditions of the validating method.**

Parameter	Value
Column	Hypersil BDS C18 (100 mm x 2.1 mm, 1.7 $\mu$ m)
Mobile Phase	Methanol and Acetonitrile in the ratio of 45:55 %v/v
Flow rate	1.2 mL/min
Run time	08 Min.
Column Temperature	Maintained at ambient temperature
Injection volume	20 $\mu$ L
Detection wavelength	238nm
Diluent	Mobile Phase

### Preparation of Standard Stock Solution

#### Preparation of Diluent

In order to achieve the separation under the optimized conditions after experimental trials that can be summarized. Stationary phase like Hypersil BDS C18 (100 mm x 2.1 mm, 1.7  $\mu$ m) column was most suitable one, since it produced symmetrical peaks with high resolution and a very good sensitivity and with good resolution. The flow rate was maintained 1.2 mL min<sup>-1</sup> shows good resolution. The PDA detector response of Triamcinolone was studied and the best wavelength was found to be 238 nm showing highest sensitivity.

The mixture of two solutions methanol and acetonitrile in the ratio of 45:55% v/v. Finally, the pH was adjusted to 10.5 with triethylamine. with gradient programming was used as mobile phase at 1.2mL/min was found to be an appropriate mobile phase for separation of Triamcinolone. The column was maintained at ambient temperature.

#### Preparation of internal standard solution

Weighed accurately about 10 mg of prednisolone working standard and transfer to 100 ml volumetric flask, add 50 ml of mobile phase and sonicate to dissolve it completely and then volume was made up to the mark with mobile phase to get 100  $\mu$ g/ml of standard stock solution of working standard. Then it was ultrasonicated for 10 minutes and filtered through 0.20  $\mu$  membrane filter.

#### Preparation of Triamcinolone standard solution

Weighed accurately about 10 mg of Triamcinolone and transfer to 100 ml volumetric flask, add 50 ml of mobile phase and sonicate to dissolve it completely and then volume was made up to the mark with mobile phase to get 100  $\mu$ g/ml of standard stock solution of working standard. Then it was ultrasonicated for 10 minutes and filtered through 0.20  $\mu$  membrane filter. Linearity was determined in the range of 8- 40  $\mu$ g mL<sup>-1</sup>.

### Stability Indicating Studies

Stability Indicating studies like acid hydrolysis, basic hydrolysis, dry heat degradation, wet heat degradation and oxidative degradation were carried out.

## 3. RESULTS AND DISCUSSIONS

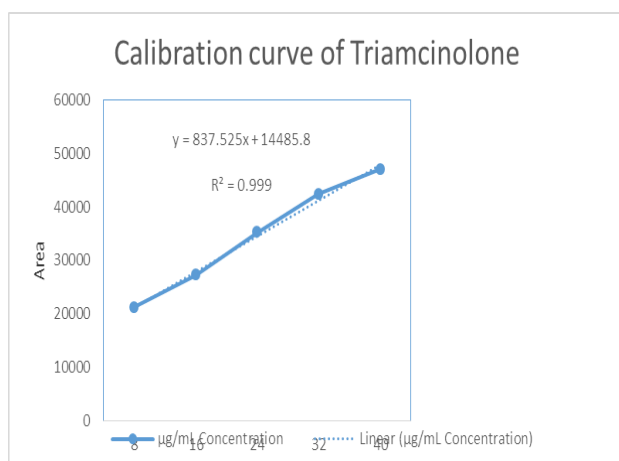
### Validation

The analytical method was validated with respect to parameters such as linearity, precision, specificity and accuracy, limit of detection (LOD), limit of quantitation (LOQ) and robustness in compliance with ICH guidelines.

### Linearity and Range

The linearity of an analytical procedure is the ability to obtain test results that are directly proportional to the concentration of an analyte in the sample.

The calibration curve showed good linearity in the range of 8 – 40 µg/ml, for Triamcinolone(API) with correlation coefficient ( $r^2$ ) of 0.9966. A typical calibration curve has the regression equation of  $y = 837.525x + 14485.8$  for Triamcinolone. Results are given in Table 2.



### Limit of Detection (LOD) and Limit of Quantitation (LOQ)

The LOD and LOQ of Triamcinolone were calculated by mathematical equation.  $LOD = 3.3 \times \text{standard deviation} \div \text{slope}$  and  $LOQ = 10 \times \text{standard deviation} \div \text{slope}$ . The LOQ of Triamcinolone was found to be 2.5535(µg/ml) and the LOD of Triamcinolone was found to be 0.8426(µg/ml). Results are given in Table 2.

**Table 2: Summary of validation parameters for the proposed method.**

PARAMETER	Triamcinolone
Linearity	8 – 40 µg/ml
Intercept (c)	14485.8
Slope (m)	837.525
Correlation coefficient	0.9966
LOD	2.5535(µg/ml)
LOQ	0.8426(µg/ml)

### Precision

The Precision of the method was studied in terms of intraday and interday precision of sample injections (16 µg/ml). Intraday precision was investigated by injecting six replicate samples of each of the sample on the same day. The % RSD was found to be 0.20%. Interday precision was assessed by analysis of the 6 solutions on three consecutive days. The % RSD obtained was found to be 0.61%. Low % RSD values indicate that the method is precise. The results are given in table 3.

### Accuracy

To study the accuracy of method, recovery studies were carried out by spiking of standard drug solution to pre-analyzed sample at three different levels i.e., at 50, 100, and 150%. The resultant solutions were then reanalyzed by the proposed method. At each level of the amount, six determinations were performed. From the data obtained, the method was found to be accurate. The % recovery and %RSD were calculated and presented in Table 4.

### Robustness

Small deliberate changes in chromatographic conditions such as change in temperature ( $\pm 2^\circ\text{C}$ ), flow rate ( $\pm 0.1\text{ml/min}$ ) and wavelength of detection ( $\pm 2\text{nm}$ ) were studied to determine the robustness of the method. The results were in favor of (% RSD < 2%) the developed UPLC method for the analysis of Triamcinolone. The results are given in table 5.

**Table 3: Results of Precision Studies.**

Triamcinolone			
Precision Studies			
Parameter	Peak Area	% RSD	% LC
Intraday precision	26856	0.50%	98.88%
	27104		99.79%
	27067		99.66%
Inter day precision	26838	0.61%	98.81%
	27147		99.95%
	27087		99.73%
Instrument:1 Acquity UPLC Waters, 2695H	27111	0.03%	99.82%
	27126		99.87%
	27109		99.81%
Instrument:2 Agilent Technologies, 1290	27117	0.04%	99.84%
	27129		99.88%
	27107		99.80%
Average			<b>99.65</b>
Std.Dev			<b>0.38</b>
% RSD			<b>0.39%</b>

**Table 4: Results of accuracy study.**

Triamcinolone						
Level %	Amount added (µg/ml)	Amount found (µg/ml)	% Recovery	Mean recovery (%)	Std. Dev	% RSD
50	08.06	08.02	99.50	99.25	0.7071	0.71%
100	16.12	16.09	99.81			
150	24.18	23.81	98.46			

**Table 5: Results of Robustness Studies.**

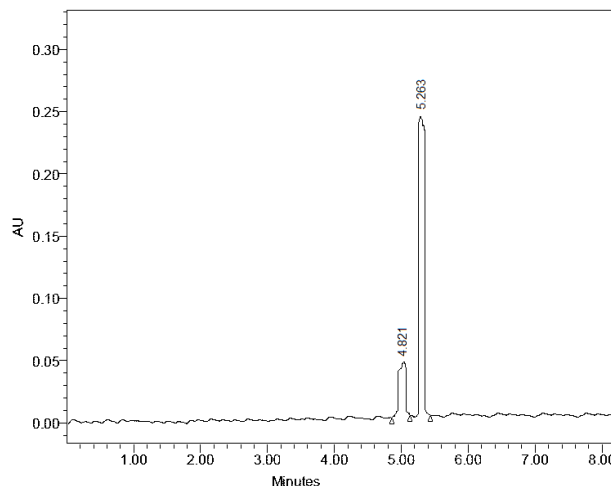
Robustness Studies			
Parameter	Value	Peak Area	% RSD
Flow Rate	Low	27553	0.11%
	Actual	27597	
	Plus	27613	
Temperature	Low	27559	0.13%
	Actual	27624	
	Plus	27620	
Wavelength	Low	27544	0.14%
	Actual	27589	
	Plus	27620	

**Results of Stability Indicating Studies**

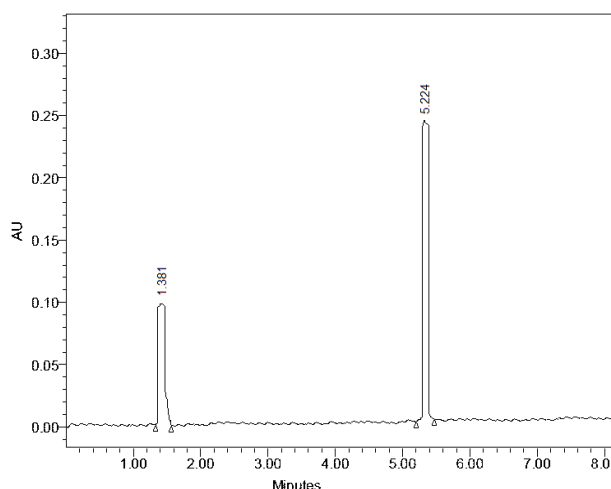
According to Singh and Bakshi, the stress testing suggests a target degradation of 20-80 % for establishing stability indicating nature of the method. UPLC study of samples obtained on stress testing of Triamcinolone under different conditions using mixture Methanol and Acetonitrile in the ratio of 45:55 %v/v as a mobile solvent system suggested the following degradation behaviour.

**a. Acid hydrolysis**

An accurate 10 ml of pure drug sample solution was transferred to a clean and dry round bottom flask (RBF). 30 ml of 0.1 N HCl was added to it. It was refluxed in a water bath at 60°C for 4 hours. Drug became soluble after reflux which was insoluble initially. Allowed to cool at room temperature. The sample was then neutralized using 2N NaOH solution and final volume of the sample was made up to 100ml with water to prepare 100ppm solution. It was injected into the UPLC system against a blank of Methanol and Acetonitrile in the ratio of 45:55 %v/v after optimizing the mobile phase composition, chromatogram was recorded and shown in Fig. 3.

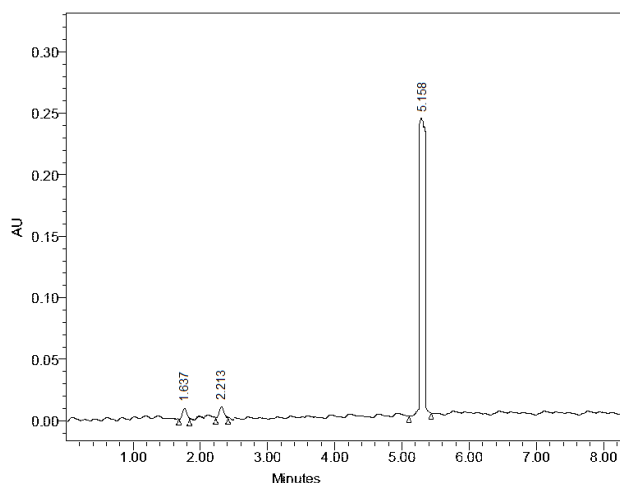
**Fig. 3: Chromatogram showing the degraded products in Acidic degradation.****b. Basic hydrolysis**

An accurate 10 ml of pure drug sample solution was transferred to a clean and dry RBF. 30 ml of 0.1N NaOH was added to it. It was refluxed in a water bath at 60°C for 4 hours. Drug became soluble after reflux which was insoluble initially. It was allowed to cool at room temperature. The sample was then neutralized using 2N HCl solution and final volume of the sample was made up to 100ml with water to prepare 100ppm solution. It was injected into the UPLC system against a blank of Methanol and Acetonitrile in the ratio of 45:55 %v/v after optimizing the mobile phase composition, chromatogram was recorded and shown in Fig. 4.

**Fig. 4: Chromatogram showing the degraded products in Basic degradation.**

### c. Wet heat degradation

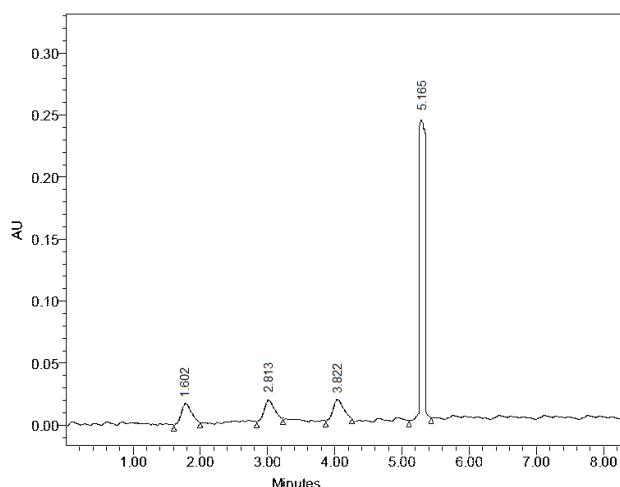
Accurate 10 ml of pure drug sample was transferred to a clean and dry RBF. 30 ml of HPLC grade water was added to it. Then, it was refluxed in a water bath at 60°C for 6 hours uninterruptedly. After the completion of reflux, the drug became soluble and the mixture of drug and water was allowed to cool at room temperature. Final volume was made up to 100 ml with HPLC grade water to prepare 100 ppm solution. It was injected into the UPLC system against a blank of Methanol and Acetonitrile in the ratio of 45:55 %v/v after optimizing the mobile phase composition, chromatogram was recorded and shown in Fig. 5.



**Fig. 5: Chromatogram showing the degraded products in Wet heat degradation.**

### d. Oxidation with (3%) H<sub>2</sub>O<sub>2</sub>

Approximately 10 ml of pure drug sample was transferred in a clean and dry 100 ml volumetric flask. 30 ml of 3% H<sub>2</sub>O<sub>2</sub> and a little methanol was added to it to make it soluble and then kept as such in dark for 24 hours. Final volume was made up to 100 ml using water to prepare 100 ppm solution. The above sample was injected into the UPLC system. The chromatogram was recorded and shown in Fig. 6.



**Fig. 6: Chromatogram showing the degraded products in H<sub>2</sub>O<sub>2</sub>**

In all degradation studies, there was a significant formation of degradation products when compared to that of a standard. This indicates that, the drug may be degraded to low molecular weight non-chromophoric compounds.

**Table 6: Stability Indicating study for the developed method.**

Nature of Stress	Degradation condition	Time (h)	Number of degradation products (Rt)
Acidic	60°C	3	1 (4.821)
Basic	60°C	9	1 (1.381)
Oxidative	RT	48	3 (1.602, 2.813, 3.822)
Wet Heat	105°C	24	2 (1.637, 2.213)

## 4. CONCLUSION

A selective and sensitive stability indicating UPLC method has been validated for the analysis of Triamcinolone in bulk drug and syrup dosage form. Based on peak purity results, obtained from the analysis of stability indicating studying samples using described method, it can be concluded that the absence of co-eluting peak along with the main peak of Triamcinolone indicated that the developed method is specific for the estimation of Triamcinolone in presence of degradation products. Further the proposed UPLC method has excellent precision, sensitivity and reproducibility. Even though no attempt has been made to identify the degraded products, proposed method can be used as stability indicating method for assay of Triamcinolone in commercial formulations.

## 5. ACKNOWLEDGEMENTS

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## 6. REFERENCES

1. International Conference on Harmonization Guideline on Validation of Analytical Procedures Text and Methodology: Q2 (R1), 2005.
2. Rania N, El-Shatieny, Fathalla F.B. Journal of Chemistry, 2015: 1-9.
3. Nguyen D.T., Guillaume D, Rudaz S, Veuthey J. L., Fast Analysis in Liquid Chromatography Using Small Particle Size and High Pressure. J Sep Sci, 2006 Aug; 29(12): 1836-48.
4. Katharina Sterz, Gerhard Scherer, Josef Ecker. A Simple and Robust UPLC-SRM/MS Method To Quantify Urinary Eicosanoids. J Lipid Res, 2013; 1-28.
5. Ashok kumar, UPLC: A preeminent technique in pharmaceutical analysis. Acta poloniae pharmaceutica- drug research, 2012; 69(30): 371-380.

6. Michael E Swartz, Ultra performance liquid chromatography UPLC: an introduction. Separation science redefined, 2005; 1: 8-14.
7. Rasheed A *et al.*, Analytical Method Development and Validation for the Simultaneous Estimation of Aspirin, Clopidogrel Bisulphate and Atorvastatin Calcium in Tablet Dosage Form. American Journal of PharmTech Research, 2014.
8. Ranjit S, Rehman Z. Current trends in stability indicating study for pharmaceutical product development. J Pharm Educ Res, 2012; 3(1): 54-63.
9. The European Pharmacopoeia convention on the Elaboration of European Pharmacopoeia, (5th Edn), 2005; 2: 2613.
10. Gupta SK, Kumar B, Sharma PK Development and validation of Spectrophotometric method for estimation of Triamcinolone in solid dosage form. Asian J Pharm Ana, 2013; 3(2): 42-43.
11. Dave BR, Jiladia MA, Vidysagar G. Simultaneous estimation of Chlorhexidine and Triamcinolone by HPLC. Int Jou of Pharm Sci, 2012; 3(4): 2850-2860.
12. Emirhan N, Filiz S, Umut A, Nursabah B. Determination of Triamcinolone in subretinal fluid by a validated HPLC method. Hacettepe University, Journal of Faculty of Pharmacy, 2005; 25(2): 71-78.
13. Aquino GDA, Stopilha RT, Pedrosa KSCR, et al. Validation of quantitative analysis method for Triamcinolone in ternary complexes by UV -Vis spectrophotometry. Rev Cienc Farm Basica Apl, 2011; 32(1): 35- 40.
14. Silva AA, Formariz TP, Scarpa MV, Oliveria AG Development and validation of HPLC method for quantitative analysis of Triamcinolone in biodegradable microparticules. Rev Cienc Farm Basica Apl, 2006; 27(2): 119-126.
15. S. Singh and M. Bakshi. Guidance on conduct of stress tests to determine inherent stability of drugs. Pharm. Technol, 2000; 26: 24-31.